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Percent means per 100 parts, where for solutions, part refers to a measure of mass

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(μ g, mg, g, kg, etc.) or volume (μ L, mL, L, etc.). In percent solutions, the amount (weight or volume) of a solute is expressed as a percentage of the total solution weight or volume.

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The percentage of solute in a solution can more easily be determined by volume when the solute and solvent are both liquids. The volume of the solute divided by the volume of solution expressed as a percent yields the percent by volume of the solution. If a solution is made by adding 40 mL of ethanol to 20 mL of

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water, the percent by volume is:

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ethanol to 20 mL of water, the percent by volume is: (16.7.4) Percent by volume = volume of solute / volume of solution \times 100 % (16.7.5) = 40 mL ethanol / 240 mL solution \times 100 % (16.7.6) = 16.7 % ethanol

16.7: Percent Solutions - Chemistry

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THE SIXTEEN-PERCENT SOLUTION: CRITICAL VOLUME FRACTION FOR PERCOLATION
RICHARD ZALLEN Department of Physics, Virginia Tech Blacksburg, VA 24061
INTRODUCTION The English call it “ value for money ” (vfm). The

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American equivalent is “ bang for the buck ” . The idea is simple: to provide a rough measure of the ratio of benefit to cost. For

THE SIXTEEN-PERCENT SOLUTION: RICHARD ZALLEN INTRODUCTION

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Solution – During this problem the equation should be rearranged to unravel for the grams of solute. We have,

Percentage of mass = (Solute ' s mass / solution ' mass) x 100%. Gram of solute = (mass percent X gram of solution)/100.

Gram of solute = (6.15 X 285) / 100 = 17.52 grams. Therefore, the gram of solute

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is 17.52 grams.

Mass Percent Formula: Meaning, Formula, Solved Examples

(b) Percentage by volume, v/v. For liquid solute, the concentration of the solution can be expressed in terms of volume percent. Example: 1000 mL of a 7.3% by

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volume of a solution of ethanol in water. It means in 100 mL solution, ethanol is there 7.3 mL. Therefore, in a 1000 mL solution ethanol = 73 mL.

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