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$$\frac{1}{(1-h)} - (A/B) \left[ \frac{h}{(1+h)} \right]$$

where  $h = b/V = B P / Z$ ,  $B = b/RT$ ,  $A/B = a / b RT^{1.5}$ , and  $Z$ , the compressibility factor, is equal to  $P V / RT$ . Calculation Procedure 1. Calculate the compressibility factor  $Z$ .

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PROPERTIES 1.5 where  $T_c$ ,  $P_c$ ,  $V_c$ ,  
 $Z_c$  are critical  
temperature, critical pressure,  
critical volume, and critical  
compressibility factor,

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respectively. Thus, for p-xylene,  $T_c =$

$$T_c =$$

$$412.3[0.567 + 0.106 - (0.106)^2]^{-1}$$

$$= 623.0\text{K}(661.8\text{ F})(\text{literature}$$

value is 616.2 K)  $P_c =$

$$106.16(0.34 + 1.378)^{-2} =$$

$$35.97\text{atm}(3644\text{ kPa})(\text{literature}$$

value is 34.7 atm)  $V_c =$



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