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~~Chapter 11 Liquids and Intermolecular~~

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Intermolecular Forces and Boiling Points

Intermolecular forces and Boiling points

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Pressure of Gases Chapter 10 (Gases) -
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Questions Chapter 11 Review Gases
Section~~

SECTION 1 Date CHAPTER 11
REVIEW Gases Class SHORT ANSWER

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Answer the following questions in the space provided.

b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripled. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOÌctbv-x

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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster

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Answers
in a gas and there is a greater distance between them.

Chapter 11 Gases Review Flashcards / Quizlet

Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe

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Answers calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

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Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon.

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*Chapter 11 Gases - An Introduction to
Chemistry*

CHAPTER 11 REVIEW Gases SECTION
1 SHORT ANSWER Answer the
following questions in the space provided.

1. b Pressure surface area. For a
constant force, when the surface area is

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- Answers
1. If the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged.
2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m²
- 3.

mc06se cFMsr i-vi - Ed W. Clark High School

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Start studying Chapter 11- Gases: Section 1: Gases and Pressure. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Chapter 11- Gases: Section 1: Gases and Pressure ...

this theory explains some of the properties

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of ideal gases. In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail. This includes the relationship among the temperature, pressure, volume, and amount of gas in a sample. SECTION 11.1

Key Terms pressure newton barometer millimeters of mercury

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*SECTION 11.1 Gases and Pressure -
Pickford High School*

CHAPTER 11 REVIEW Gases SECTION
2 SHORT ANSWER Answer the
following questions in the space provided.

1. State whether the pressure of a fixed
mass of gas will increase, decrease, or stay

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Answers
the same in the following circumstances:
increase a. temperature increases, volume
stays the same decrease b. volume
increases, temperature stays the same

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Modern Chemistry 93 Gases CHAPTER
11 REVIEW Gases SECTION 1 SHORT
ANSWER Answer the following questions
in the space provided. 1. _____ Pressure =
. For a constant force, when the surface
area is tripled the pressure is (a) doubled.
(b) a third as much. (c) tripled. (d)
unchanged. 2. _____ Rank the following

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Answers in increasing order.

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